Hydrogen sulfide plasma passivation of gallium arsenide

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Improvement in the electrical properties of the GaAs surface has been accomplished using a room-temperature hydrogen sulfide plasma. The surface has then been protected by a $300\,^{\circ}\text{C}$ plasma enhanced chemical vapor deposition (PECVD) SiO_2 film. This treatment is highly reproducible due to computer control of process parameters and long-lasting due to the SiO_2 cap. Improved C-V characteristics were observed, showing interface trap densities in the high 10^{11} cm $^{-2}$ eV $^{-1}$ range. Photoluminescence (PL) measurements on the sulfided samples showed increased intensity over the untreated samples.

The poor quality of the native oxide of GaAs has made surface passivation extremely challenging and necessitated the use of deposited insulators. Unlike the well-behaved Si-SiO₂ interface, the high density of surface states at the dielectric-GaAs interface results in Fermi-level pinning¹ which inhibits device performance.^{2,3} A well passivated surface with a low interface state density is desirable in the development of a viable metal-insulator-semiconductor (MIS) technology, as well as for the passivation of other III-V devices, such as heterojunction bipolar transistors (HBT).

Recently, there have been many reports of dramatic improvements using sulfur as a surface treatment. The original application method was a spin-on Na₂S solution, ^{4,5} but (NH₄)₂S applied through immersion has shown better performance by being more durable and exhibiting less leakage current in *p-n* diodes. ⁶⁻⁸ While there have been remarkable improvements in surface recombination velocity ⁵ and HBT performance, ^{4,9} there have been some problems as well, such as poor reproducibility and aging.

In this letter we will describe a new technique using a room-temperature H₂S plasma followed by a higher temperature (300 °C) plasma enhanced chemical vapor deposition (PECVD) SiO₂ film. There are several advantages to plasma sulfidation over conventional wet chemical treatments. First, there is a high degree of reproducibility, since all the process parameters such as gas flow, chamber pressure, treatment time, and rf power are computer controlled. Second, the sulfidation and subsequent SiO₂ deposition are performed in the same chamber, so there is no exposure to air between the two treatments. Third, this is a long lasting treatment, since the SiO₂ cap prevents air degradation of the sulfide layer.

The samples used in this study were commercially available MOCVD grown n-type (100) GaAs epitaxial layers 2 μ m thick doped to 1×10^{15} on n^+ substrates. They were degreased in hot TCA, acetone, IPA, and blown dry in N₂, then etched in HCl:H₂O (1:1) to remove the native oxides, rinsed in DI water and blown dry in N₂. After etching they were promptly loaded into the PECVD chamber which was immediately pumped down and purged with argon. The PECVD reactor is a parallel plate SEMI group model MPB 1000 operating at 13.56 MHz. It is a direct

plasma configuration, with the samples exposed to the plasma.

The room-temperature H₂S plasma treatments were performed at 200 mTorr for durations of 1-20 min. The temperature was then raised to 300 °C under an argon flow and 100 nm of high quality SiO₂ (Ref. 10) were deposited from silane, nitrous oxide, and helium using a process described elsewhere. 11 The longer H₂S treatment times should not result in thicker sulfide layers, since excess sulfur on the wafer should sublime away as the temperature was raised above 110 °C. 12 However, the longer sulfidation times did result in thinner SiO2 films, as shown in Fig. 1. We believe this is due to excess sulfur remaining on the unheated chamber walls, which then consumed reactant gases and introduced a delay in SiO2 formation. Similar problems were observed when high-temperature (300 °C) sulfide treatments were attempted. Two minutes was selected as the sulfidation time, as this sufficiently coated the wafer with sulfur without unduly affecting the SiO₂ film.

Samples for C-V measurement were patterned on the front with aluminum pads and metallized on the back with a Ni/Ge/Au ohmic contact. Results of high-frequency C-V measurements are shown in Fig. 2. The control sample exhibits peculiar kinks in the data, while the sulfided sample shows a very smooth and continuous plot, although with significant hysteresis. A Terman analysis¹³ of the sulfided sample's forward curve yields a mid-gap trap density

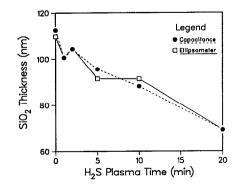
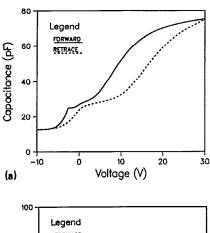
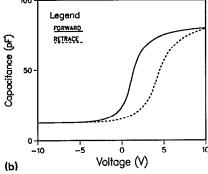


FIG. 1. Variation of SiO_2 film thickness with H_2S plasma treatment time, as measured with single wavelength ellipsometry (632.8 nm) and oxide capacitance.





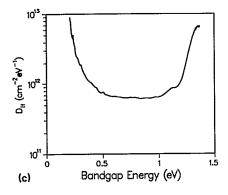


FIG. 2. C-V characteristics of SiO₂ on GaAs. (a) Control sample with no H₂S treatment, (b) sulfided sample with 2 min H₂S plasma treatment, (c) trap distribution of the sulfided sample's forward curve calculated using

in the high 10^{11} cm⁻² eV⁻¹ range. There are several possible explanations for the hysteresis, such as reoxidation of the surface from the reactive oxygen of the SiO₂ film, charge trapping in the sulfide compounds at the interface, or SiO₂ defects produced by initial reactions with the sulfide layer. Attempts to reduce the hysteresis with an intermediate amorphous silicon layer between the sulfide and the SiO₂ proved unsuccessful, as the SiO₂ films would bubble rapidly upon subsequent heat treatment.

Photoluminescence (PL) measurements were taken at 20 K using the 514.5 nm line of an argon ion laser operating at 0.2 W/cm². Three samples were examined, a bare GaAs sample, a SiO2 on GaAs sample, and a sample with a 2 min sulfide treatment followed by a SiO2 cap. The SiO2 layers were thin (approximately 30 nm) for the PL study.

The PL analysis consisted of comparing the peak intensity of the conduction band to valence band transition (located at 820.7 nm), generally considered to scale inversely with surface state density. 14 The peak intensity for the SiO₂ on GaAs sample was slightly lower than the bare control sample, but the signal from the SiO2 capped sulfided GaAs sample was nearly twice as strong as the SiO2 on GaAs sample and stronger even than the bare sample.

In conclusion, a promising technique has been demonstrated for sulfidation of the GaAs surface that is both reproducible and long lasting. Improvements in the MIS C-V characteristics were observed, although significant hysteresis remains. Further work will be required to reduce this hysteresis, and to determine the applicability of this approach to other III-V semiconductor surfaces.

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